

Iodine content in the produced and marketed iodized edible salt in Semnan providence, the primary salt-producing province in Iran

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Abstract Background: Iodine deficiency represents a significant global public health issue, as it can result in Iodine Deficiency Disorders. Worldwide initiatives have diminished the incidence of these disorders in numerous areas via Universal Salt Iodization. The purpose of this study was to evaluate the iodine content in salt samples manufactured and distributed in Semnan Province, Iran.

Methods: A total of 373 salt products, available in markets and produced by factories in the Semnan province of Iran, were collected. The samples were analyzed to determine iodine concentration using a titration instrument, in accordance with the standards of the Iranian National Standardization Organization No. 3696 and Codex Alimentarius.

Results: Iodine was found in 86.6% of edible iodized salt samples at a standard concentration, with a mean \pm SD of 40.93 ± 12.07 mg/kg. The iodine concentration ranged from a minimum of zero to a maximum of 90.71 mg/kg.

Conclusions: The article primarily indicated that the majority of salts produced and marketed in Semnan province contain iodine concentrations within the standard range.

Keywords: Iodized salts, Iodine content, Semnan, Iran.

I. Introduction

Iodine, a vital micronutrient, is key to the proper development of the central nervous system and thyroid function [1]. It regulates essential bodily functions including temperature, growth, and reproduction [1]. Iodine is found in various chemical forms, primarily as iodide (the ion form of iodine) and iodate (a salt or ester of iodic acid) [2]. Iodine deficiency presents a significant global public health concern due to its potential to cause Iodine Deficiency Disorders (IDD), including impaired cognition in children, cancer, and developmental disease [3-5]. excessive iodine intake also lead to thyroid disorder and cancer [5]. Methods such as incorporation of iodine in salt have effectively addressed iodine deficiency, leading to global progress in iodine intake [4, 6]. Global efforts have decreased the prevalence of iodine deficiency disorders in numerous regions, through Universal Salt Iodization (USI) [7]. Some studies have shown that Iran has successfully USI-implemented a 29-year national salt iodization program, resulting in sufficient iodine intake among the population [8-10]. More than 98% of households consume iodized salt in Iran, meeting the criteria set by the World Health Organization (WHO) [9]. The general population in Iran has sufficient iodine levels, leading to Iran declaring itself iodine deficiency disorder-free in 2000 [8, 10]. The prevalence of goiter decreased significantly after 17 years of iodization in Iran [11]. Some provinces lack adequate iodine nutrition and require special attention in Iran [12]. It should mention, both lack of iodine and excessive iodine intake can cause health problems so periodic monitoring of population iodine status is important for ensuring sufficient intakes [7]. Given the importance of continuous measurement of iodine concentration in salt, this study aimed to evaluate iodine concentration in the produced and marketed salt samples in Semnan Province, the primary province for salt production in Iran.

II. Materials and Methods

2.1. Sampling

A total of 373 samples were randomly collected from five cities in Semnan province, namely Semnan, Garmsar, Damghan, Mahdishahr, and Sorkheh, between March 2020 and March 2024. These samples were all sent to the Food and Drug Control Laboratory at Semnan University of Medical Sciences. Of these, 106 samples were purchased from markets, while 267 were acquired from Salt Production Factories.

2.2. Reagents

We used three reagents as follows: 1- Titrant: sodium thiosulfate c(Na₂S₂O₃) 0.01 mol/L (0.02 normal) 2- Sulfuric acid 2 Normal 3- Potassium iodide 10% w/v.

2.3. Sample preparation

Similar to previous research [13], all samples were analyzed for iodine concentration at the Food and Drug Control laboratory, following the guidelines of the Iranian National Standardization Organization (INSO) No.3696, the Combined Compendium of Food Additives Specifications [14], and the Codex Alimentarius monograph [15].

12.5-gram sample was weighted out into a breaker and dissolved in approx. 65 mL of distilled water. Then, 1 mL of sulfuric acid (2N) and 5 mL of potassium iodide solution (10%) were added and allowed to react for 2 min under stirring, turning the solution yellow. The breaker was sealed and placed in a dark area for 10 minutes.

2.4. Analysis and calculations

The iodine released in the solution was titrated after the initial point using sodium thiosulfate (0.02 N) through automated titration methods (Titrand 906, Metrohm, Switzerland) and was managed with Taimo software (version 2). Titrand comprised three instruments: 1- Pt Titrode 6.0431.100 2- Electrode cable 6.2014.020 3- Dosing unit 6.3032.2020.

The total iodine content is expressed in mg/kg (ppm). One milliliter of 0.01 mol/L c(Na₂S₂O₃) is equivalent to 0.2115 mg of iodine. The final equation is presented below:

$$\text{total iodine (ppm)} = EP1 \times 0.2115 \times 1000 \times \frac{\text{Titter}}{\text{sample size}}$$

In this calculation formula, EP1 represents the milliliters of c(Na₂S₂O₂) up to the endpoint, the sample size is the weight of the sample in grams (12.5 gr), 1000 is for 1 kg, and Titer is the titer number of the c(Na₂S₂O₂) solution.

2.5. Method validation procedures

All of the samples were analyzed with the validated automated titrator techniques which fitted with a platinum electrode was applied. The iodine detection method was pre-validated at the Food and Drug Control laboratory. Linearity of the method was assessed using a graph plotting iodine concentrations. The graph depicted analyte concentrations at five levels (0, 15, 30, 60, 120 mg/kg), each with three replicates (Fig. 3). A regression line was drawn, yielding a correlation coefficient ($R^2 = 0.9990$) through the least squares method, to evaluate the relationship between the actual area and the measured iodine concentration.

2.6. Statical analysis

Quantitative data are presented as mean \pm SEM, and qualitative data are expressed as percentages (%). All analyses were performed using Prism 9 GraphPad software. Differences in quantitative data between groups were assessed using one-way ANOVA followed by Tukey's post-hoc test. A two-tailed p-value of less than 0.05 was considered statistically significant.

III. Results

The current study utilized titrator techniques to detect iodine levels in iodized salt samples. The results, as presented in Table 1, indicate that the majority of samples fell within the standard iodine range (Mean \pm SD; 40.93 \pm 12.07). Specifically, 86.6% of the samples contained iodine within the standard range (30-60 mg/kg), while the remaining 13.4% had iodine levels outside the standard range. The iodine concentration ranged from a minimum of zero to a maximum of 90.71 mg/kg.

Table 1; Iodine detected in produced and marketed edible iodized salt samples

Parameters	Origine of samples		Total (precent)
	Markets	Salt Production Factories	
Number of Samples	106	267	373
Mean Iodine Content \pm SD (mg/kg)	39.64 \pm 13.88	41.45 \pm 11.26	40.93 \pm 12.07
Samples with Iodine Content < 30 mg/kg	16 (15%)	25 (9%)	41 (11%)
Samples with Iodine Content > 60 mg/kg	1 (1%)	8 (3%)	9 (2.4%)
Samples with Normal Iodine Content (30-60 mg/kg)	99 (84%)	234 (88%)	323 (86.6%)
Minimum Iodine Content (mg/kg)	0	3.23	0
Maximum Iodine Content (mg/kg)	80.85	90.71	90.71

No significant variations are observed in the iodine concentration of iodized salt samples across different years and seasons (refer to Fig. 1 and 2).

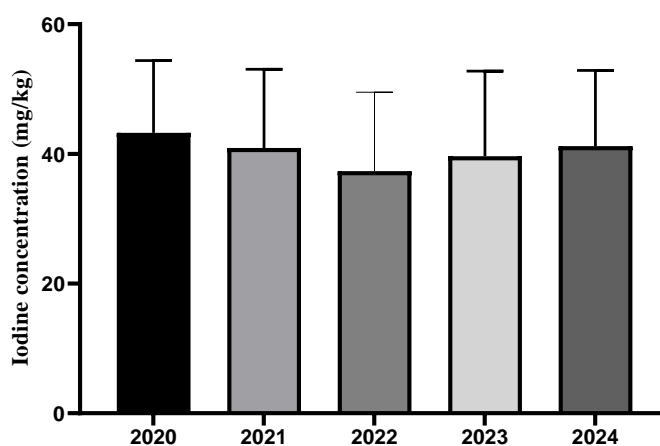


Figure 1: The iodine concentration in the iodized salt samples by years. Values are expressed as the mean \pm SEM. Data were analyzed by one-way ANOVA multiple comparison test followed by Tukey's multiple comparison tests. (p -value > 0.05)

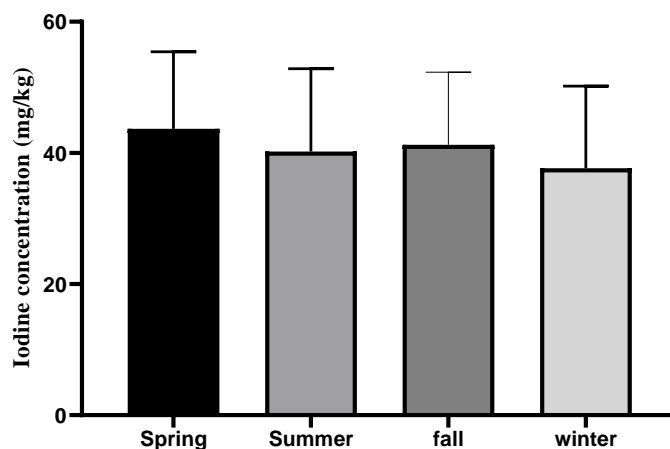


Figure 2; The iodine concentration in the iodized salt samples by seasons. Values are expressed as the mean \pm SEM. Data were analyzed by one-way ANOVA multiple comparison test followed by Tukey's multiple comparison tests. (p -value > 0.05)

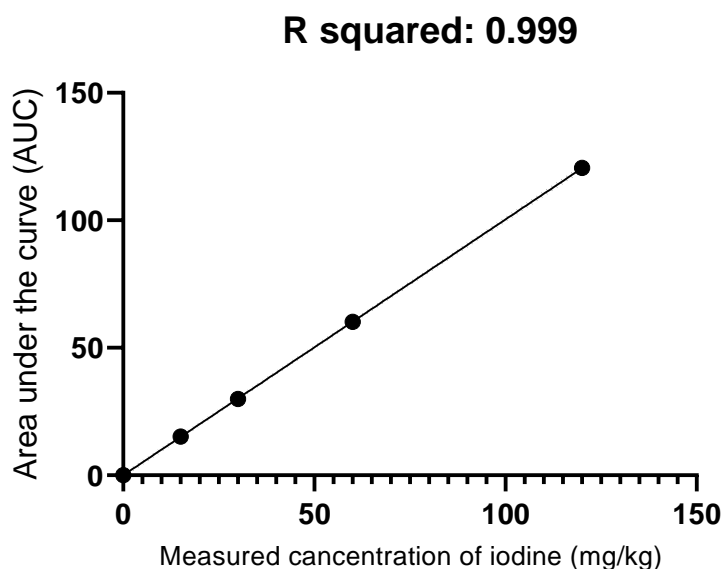


Figure 3: Linearity plot for five different concentrations of Iodine using automated titrator instrument.

IV. Discussion

The current study revealed that 13.4% of iodized salt samples were outside the standard iodine range. Specifically, 11% had iodine concentrations below the standard threshold, while 2.4% exceeded it. It is important to note that both insufficient and excessive iodine intake can lead to health issues. In Iran, studies on the monitoring of salt's iodine levels indicated that 49% had iodine concentrations beyond the standard limit (Table 2). Consequently, the iodine content in consumable iodized salt raises concerns, necessitating additional research and regulatory scrutiny.

Table 2: Condition of iodine in iodized salts in Iran

Province	Number of salt samples were out of standard range for iodine	year	reference
Markazi (Center of Iran)	50%	2020	[16]
Qom (Center of Iran)	48.33%	2017	[17]
Lorestan (West of Iran)	35.41%	2015	[13]
Kermanshah (Center of Iran)	50%	2013-2014	[18]
Khorasan-e-Razavi (East of Iran)	32.8%	2010-2015	[19]
Tehran (Center of Iran)	56.08%	2010	[20]
Tabriz (North of Iran)	44.1%	2006	[21]
Semnan (Center of Iran)	31.2%	2004	[22]

Bushehr (South of Iran)	89%	2000	[23]
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Iran and some countries like Croatia have successfully implemented USP, emphasizing the importance of monitoring iodine intake to prevent deficiencies and excesses [8, 10, 24]. Studies indicate that Iranian salt intake remains high (9.52 g/day, double WHO recommendation) [21, 25-27]. One possible reason for the success of the national USP, despite the existence of non-standard iodine levels in some salts, could be the high consumption of salt by Iranians. This high consumption, while potentially harmful, may help compensate for the lack of iodine in iodized salt. However, challenges persist in ensuring sufficient iodine levels, particularly during pregnancy and in young people [6, 28]. Iodine supplements can improve fetal neurodevelopment, but their effect on maternal thyroid function is inconclusive [29, 30]. In light of the findings, conducting further research and establishing a serious audit for producing edible iodized salt is recommended.

V. Conclusion

This article mainly presented that most of produced and marketed salts in Semnan province has iodine concentration in standard range.

Ethical Considerations: All ethical principles were considered in this article.

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Conflict of interest: The authors declared no conflicts of interest.

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